Full Marks-70

Cradite.5

# Unit-I (A) Alkyls and aryls transition metals

Types, routes of synthesis, stability and decomposition pathways,

Organocopper in organic synthesis

# (B) Compounds of transition metal-carbon multiple bonds.

Alkylidenes, alkylidynes, low valent carbens and carbynes symbesis, nature of bond, structural characteristics, Nucleophilic and electrophilic reactions the ligands, Roles in organic symbenses. Flaxional organometallic compounds, Fluxionality and dynamic equilibrial unit-it Transition metal = complexes.

#### Unit-II 7

Transition metal a complexes with unsaturated organic molecules alkenes, alkynes, alkyl, diene, dienyl, arene trienyl complexes, their structural features and important nucleophilic and electrophilic reactions.

## Unit-III Homogeneous Catalysis.

Stoichiometric reactions for catalysis, homogeneous catalytic hydrogenation, Zeigher Natta polymerization of olefins, catalytic reactions involving (O, [e.g. hydro-carbonylation of olefins, (no reaction)], coopalidation reactions, activation of C-H bond.

## Unit-IV (A)SupramolecularChemistry

Introduction, Non covalent interactions, self-assembly in supromolecular chemistry, Reactivity and catalysis design and synthesis, transport processes and carrier design, superamolecular devices.

# (B) Photo chemistry of metal complexes.

Basis of photochemistry, properties of excited states, excited states of metal complexes and their comparison with organic compounds.

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Photo-substitution, photo-oxidation and photo-reduction, Excited electron transfer, Reactions of 2, 2-bipyridines and 1, 10 phenanthroline complexes, metal complexes sensitizers, Application of photochemical reactions of co-ordinance compounds.

#### Unit-V (A) Molecular rearrangement

D and A process, reactions of geometrical and optical isomers, optical inversions, isomerisation and recemisation of octahedral

complexes, intermolecular and intramolecular rearrangement.

(B) Spectroscopic Application: Application of Mossbauer and ESR spectroscopy in elucidation of structure of inorganic molecule.

## Books Recommend:

- 1. Organometallic Chemistry-Ayodhya Singh and Ratnesh Singh
- Organometallie Chemisty- RC Mehrotr and A. Singh
- Organometalile Chemistry of transition metals- Robert H. Crabtree
- 4. Organometallic Compounds-Indrajeet Kumar.
- 5. Supramolecular chemistry-concept and perspective- J.M. Lehn
- 6. Introduction to Supramolecular chemistry- Hiclena- Dodziuk
- 7. Supramolecular chemistry Norendra N. Ghosh.
  - Photochemistry- Carle E. Wayne and Richard P. Wayne
- 9. Inorganic chemistry- Gary Walfsberg
- 10. Inorganic ehemistry- J. E. Hulhey, A. Keiler, L. Keiler, D.K. Medhi
- 11. Inorganic Chemistry -G.L. Miessier and D.A. Tarr
- 12. Advanced Inorganic chemistry -Cotton and Wilkinson T.



#### Semester-IV Elective Course-1b Physical Chemistry Special

# Full Marks-70

Credits-5

## Unit-I (A) Hartree Fock Theory:

Born oppenheimer approximation. Salter-Condon rule, Hartree-Fock equation, Koopman theory.

## (B) Semi Empirical Theories

HMO Theory of  $\pi$  systems. Bond order, Free valence and charge density, and its calculation. Extended Huckle theory,

### Unit-II Catalysis and Oscillatory Behaviour

Kinetics of catalytic reaction, Arrhenius intermediates, vant-Half intermediates, Theory of acid-base catalyst, Bronsted catalysis law, Hommet agreation Oscillatory reactions.

### Unit-III (A) Kinetics of condensed phase Reaction.

Factors determining reaction rate in solution. Transition state theory in solution, kinetics of sonic reaction. Dependence of rate constant on ionic strength and dielectric constant of the medium. Bronsted Bjerrum equation.

#### (B) Study of Fast reactions.

Flash Photolysis, relaxation techniques, Molecular beam and shock Tube kinetics, stop flow method.

### Unit- IV Kinetics of Electrode reactions

Faradic and non-faradic current rate law in faradic process, current density, factors affecting electrode-reaction, Effect of double layer structure on electrode reaction rates.



#### Unit-V (A) Corrosion

Scope and economic of corrosion, causes and types of corrosion, electrochemical theories of corrosion Method of protection the corresion

#### (B) Thermodynamics of solids

Specific heat of solids, Einstein heat capacity equation Debye theory of specific heat.

### Books Suggested.

1. Physical chemistry

2. Advance Physical chemistry

Chemical kinetics
 Introduction to chemical
 Thermodynamics

P.W. Atkins

: Gurdeep Raj : Keith, J. Laidler

R.P.Rastogi & R.R. Mishra

#### Semester-IV Elective Course-1c Organic Chemistry Special

# Full Marks-70

Credits-5

#### Unit-I Terpennoids

Introduction, classification, isoprene rule and special isoprene rule. Structural determination, stereochemistry and synthesis of citral,  $\alpha$ -Terpeniol, camphor, santonin

#### Unit-II Alkaloids

Introduction, classification, general method of structure determination.

Structure and synthesis of the following compounds- Papaverine,
Nicotine, Atropine and Morphine.

#### Unit-III Drug Design

- (a) Introduction, classification of drugs. Development of new drugs. Procedures followed in drug design. Structure activity relationship. Receptor. Theories of drug activity with emphasis on Drug-receptors interactions.
  (b) Application of Mass. IR. IIP-Visible NMR CH & □C in elacidation
- (b) Application of Mass, IR, UV-Visible, NMR (<sup>1</sup>H & <sup>13</sup>C) in elucidation of structure of organic molecules.

#### it-IV Drugs

- Antineoplastic Agents: Introduction, Cancer chemotherapy, role of allylating agents, antimetabolites, natural products and hormones in treatement of cancer. Synthesis of mechlorethamine, cyclophosphamide, uracil-mustards, 6-mercaptopurine, melphalan.
  - 2. Cardiovascular Drugs: Cardiovascular disease, drug inhibition of peripheral sympathetic function, direct acting arteriolar dilators. Synthesis of amyl nitrate, hydrolaxine verapamil, diazoxide propanol,
- Synthesis of amyf nitrate, hydrolaxine verapamil, diazoxide propanol, sorbitrate, quinidine, Methyldopa, atenolol and oxyprenolol.

  3. Anti-tubercular Drugs: PAS, Isoniazid, Ethambutol Thiosemicarbozone,



### Unit-V Heterocyclic Compounds

- 1 Renzfused five membered heterocyclic compounds: Classification, nomenclature of aromatic heteroatoms: Synthesis and reaction of benzopyrole, benzofuran, benzothiophene
- 2. Five and Six membered Heterocycles1 with two or more heteroatoms: Synthesis and reaction of oxazole, isooxazole, pyrazole. Imidazole thiazole diazine and tetrazines.
- 3. Seven and large membered Heterocycles with two or more heteroatoms: Synthesis and reaction of azenines, oxenines, diazenines, azocines and thiapines.

### Books Recommend

- 1. Natural Products-Chemistry and Biological Significance by J. Mann, R.S. Davidson, I.B. Hobbs, D.V. Banthrope and J.B. Harborne.
- 2. Organic Chemistry by LL. Finar.
- 3. Rodds Chemistry of Carbon Compounds by S. Coffey.
- 4. Natural Products Chemistry by Jagdamba Singh and Java singh. 5. The Chemistry of Natural Products by P.S. Kalsi
- 6. Chemistry of Natural Products by Nakamshi.
- 7. An Introduction to Medicinal Chemistry by Graham L. Patrick. 8. Textbook of Organie Medicinal and armaceutical Chemistry by Charles O. Wilson, Ole Gisvold& Robert F. Doerge.
- 9. Principles of Medicinal Chemistry by Wilam 0. Foye, Thomas L. Lemice and David A. Williams.
- 10 Rurgers Medicinal Chemistry and Drug Discovery by M.E. Wolft
- 11. Hetcrocylie Chemistry by RR. Gupta, M. Kumar and V.Gupta.
- 12. Heteroeyclic Chemistry by T.L. Gilchrist. 13. Organic Chemistry by LL, Finar,



Semester-IV

Full Marks - 50	Elective Course (P) 2a Practical (Inorganic Chemistry Special Duration of Exam 12 hrs.

Credit - 5 1. Qualitative analysis of Inorganic mixture containing six radicals including Mo, V, W, Ce

2. Analysis of atleast two metal ions in alloys and minerals (a) Dolomite (b) Brass (C) Solder (d) Bauxite

OR 196% Spectrophotometric determination of Fe, Ni, Ma, Cr, V, Ti, F, NO3, and PO4" etc.

15

Books Recommended:

1. Qualitative Analysis - A. L. Vogel 2. Quantitative Analysis - A. I. Vogel

Semester - IV Elective Course (P) 2b Practical (Physical Chemistry Special)

Full Marks - 50 Duration of Exam 12 hrs.

Credits- 5 (Marks 30)

Two experiments have to be set.

 Couductometric titration of strong acid and strong base (NaOH+HCI) 2. Potentiometrically pH of a given solution using hydrogen electrode or

quinhydrone electrode.

3. Potentiometric Experiments Determination of Acid-base titration.

 Determination of partition coefficient of Iodine between CCl<sub>4</sub> and water. 5. Determination of partition coefficient of KI+I2 = KI4between CCI4 and

water. 6. Viva-vnce

-15 7. Note Book

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Semester - IV Elective Course (P) 20 Practical Organic chemistry (Special)

Full Marks - 50 Duration of Exam 12 hrs. Credits - 5

05 Marks

## Any two experiments have to be set (Marks 30)

- 1. Separation and identification of organic compounds using chemical methods from organic mixtures containing up to three components
- 2. Preparation of organic compounds involving several stages
- 3. Estimation of carbohydrates, protein, aminoacids, ascorbic acid, blood cholesterol and aspirin by UV - visible Spectrophtometric method.
  - 4. Vivo Vace 15 Marks
  - 5. Note Book





Semester - IV GE-1 45